

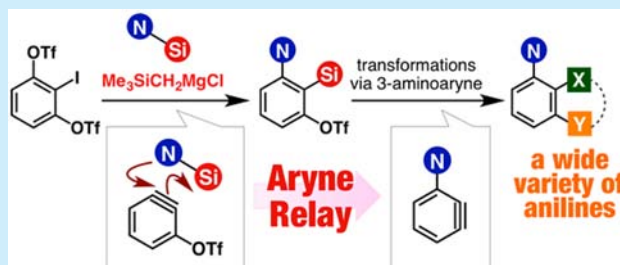
Aryne Relay Chemistry en Route to Aminoarenes: Synthesis of 3-Aminoaryne Precursors via Regioselective Silylamination of 3-(Triflyloxy)arynes

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S Supporting Information

ABSTRACT: A facile synthetic method for preparing 3-amino-2-silylaryl triflates via regioselective silylamination of 3-(triflyloxy)arynes with *N*-silylamines is described. Fluoride-mediated generation of 3-aminobenzynes from 3-amino-2-silylphenyl triflate, easily prepared by this method, in the presence of various arynophiles efficiently afforded diverse aniline derivatives, including a 5-aminocoumarin derivative, demonstrating the utility of aryne relay approach.



Aniline derivatives play an important role in several fields, including materials science and medicinal chemistry.^{1,2} A number of synthetic methods, such as the reduction of aromatic nitro compounds, nucleophilic aromatic substitution of electron-deficient aryl halides, and Buchwald–Hartwig amination, have been developed for the preparation of aromatic amines.¹ Considering the recent advances in synthetic aryne chemistry,^{3,4} amino group substituted arynes could serve as promising intermediates in the preparation of various aniline derivatives. Nevertheless, only a few studies on the transformation via aminoaryne species have been reported.⁵ Herein, we report that 3-amino-2-silylaryl triflates, which have been anticipated to serve as 3-aminoaryne precursors, are easily prepared via regioselective silylamination of 3-(triflyloxy)arynes efficiently generated from *o*-iodoaryl triflate type aryne precursors. We also show that a 3-aminobenzynes is efficiently generated from a 3-amino-2-silylphenyl triflate synthesized by this method, which rendered a diverse range of aniline derivatives easily accessible, demonstrating the synthetic utility of aryne relay chemistry.

Since a wide range of aromatic compounds can be prepared via arynes that are generated from *o*-silylaryl triflates,^{3,4} we initially aimed at preparing 3-amino-2-(trimethylsilyl)phenyl triflate **1**. We assumed that the treatment of **1** with a fluoride ion source in the presence of an arynophile would result in the generation of 3-aminobenzynes **I**, which could be used for the synthesis of a variety of aniline derivatives through various transformations such as cycloaddition, addition of a nucleophile, and direct difunctionalization (Figure 1A). However, our initial attempt to prepare 3-(dimethylamino)-2-(trimethylsilyl)phenyl triflate **1** (*R*, *R'* = Me) using the conventional method⁶ was unsuccessful (Figure 1B); while carbamate **3** was obtained from 3-(dimethylamino)phenol (**2**) in a low yield, an attempt

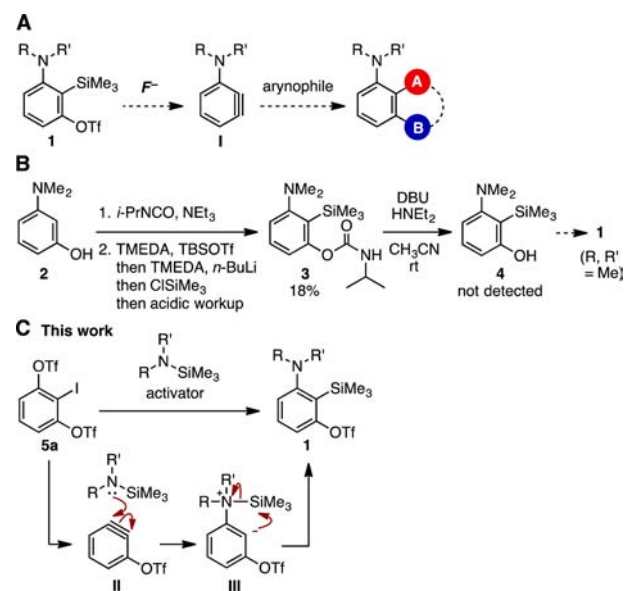


Figure 1. Proposed strategy. (A) Synthesis of aniline derivatives via 3-aminobenzynes **I** generated from 3-amino-2-silylphenyl triflate **1**. (B) Our initial attempt to prepare **1**. (C) This work: synthesis of 3-aminobenzynes precursor **1** via regioselective silylamination of 3-(triflyloxy)benzynes (**II**).

to deprotect **3** to obtain *o*-silylphenol **4** was unsuccessful, and instead, desilylprotonated phenol **2** was obtained.

We then attempted to prepare 3-amino-2-silylphenyl triflate **1** via silylamination⁷ of 3-(triflyloxy)benzynes (**II**) (Figure 1C).⁸ This idea was based on our recent studies on the chemistry of

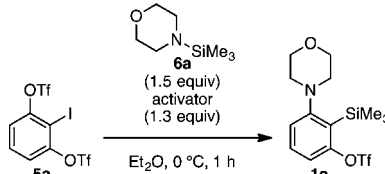
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3-(triflyloxy)arynes; we have developed an efficient method for their generation from 1,3-bis(triflyloxy)-2-iodoarenes such as **5a** and examined their reaction with various arynophiles.^{8a,f} On the basis of these studies, we assumed that the addition of the amino group of *N*-silylamine to **II** would take place regioselectively at the distal site owing to the presence of an electron-withdrawing triflyloxy group and subsequent migration of the silyl group would result in the formation of desired **1**.

Based on this scenario, we extensively screened the conditions for efficiently synthesizing 3-morpholino-2-(trimethylsilyl)phenyl triflate (**1a**) via the silylamination of 3-(triflyloxy)benzyne, generated from 1,3-bis(triflyloxy)-2-iodobenzene (**5a**), with *N*-(trimethylsilyl)morpholine (**6a**) (Table 1). Our initial attempts of treating a mixture of **5a** and **6a** in

Table 1. Optimization of the Reaction Conditions



entry	activator	yield ^a (%)
1	<i>n</i> -BuLi	27
2	<i>i</i> -PrMgCl·LiCl	33
3	PhMgBr	28
4	Me ₃ SiCH ₂ MgCl	73
5 ^b	Me ₃ SiCH ₂ MgCl	80
6 ^{b,c}	Me ₃ SiCH ₂ MgCl	81 ^d

^aYields based on ¹H NMR analysis, unless otherwise noted. ^bThe reaction was performed at room temperature. ^c**5a** (10 mmol, 5.0 g) was used. ^dIsolated yield.

diethyl ether at 0 °C with an activator, such as *n*-butyllithium, isopropylmagnesium chloride–lithium chloride complex, or phenylmagnesium bromide, afforded the desired **1a** only in low yields (entries 1–3). The use of (trimethylsilyl)methylmagnesium chloride as an activator⁹ largely increased the yield of **1a** (entry 4), which was further improved by performing the reaction at room temperature (entry 5); these conditions allowed for a gram-scale synthesis of **1a** without decreasing the yield (entry 6). The formation of undesired byproducts, such as a regioisomer or a triflone derivative, which could be formed via the thia-Fries rearrangement of an aryl anion intermediate (**III**, Figure 1C),^{8a,f} was not observed. On the other hand, an attempt to prepare **1a** from 1,3-bis(triflyloxy)-2-(trimethylsilyl)benzene (**7**),^{8b,c} another 3-(triflyloxy)benzyne precursor, according to the conditions for silylamination of arynes reported by Yoshida, Kunai, and co-workers,⁷ was unsuccessful; treating a mixture of **7** and **6a** with potassium fluoride and 18-crown-6-ether did not afford **1a**, probably due to the prior cleavage of the N–Si bond of **6a** (Scheme 1). These results indicated the advantage of using 1,3-bis(triflyloxy)-2-iodobenzene (**5a**) as a precursor of 3-(triflyloxy)benzyne and the silylmethyl Grignard reagent as its activator for efficiently achieving the silylamination of 3-(triflyloxy)benzyne without cleaving the weak N–Si bond.¹⁰

The optimized conditions were successfully employed for the preparation of various 5-substituted 3-morpholino-2-(trimethylsilyl)aryl triflates (Figure 2). For example, methyl-, *p*-anisyl-, or *p*-fluorophenyl-substituted 3-(*N*-morpholino)aryne

Scheme 1. Attempt To Prepare **1a** from an *o*-Silylphenyl Triflate Type Precursor

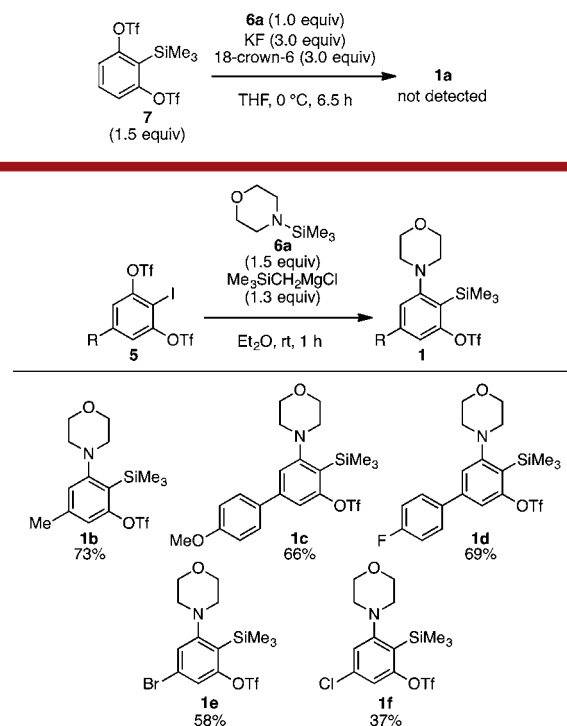


Figure 2. Synthesis of various 5-substituted 3-morpholino-2-silylaryl triflates.

precursors **1b–d** were obtained in good yields via silylamination of 3-(triflyloxy)arynes, generated from the corresponding 5-substituted 1,3-bis(triflyloxy)-2-iodobenzenes. Furthermore, the silylamination of 5-bromo- or 5-chloro-3-(triflyloxy)benzyne afforded 3-morpholinoaryne precursors **1e** and **1f** bearing a transformable halogeno group in moderate yields without affecting the halogeno groups.

A wide range of *N*-(trimethylsilyl)amines were also applicable to the silylamination of 3-(triflyloxy)benzyne, which provided various 3-amino-2-silylaryl triflates **1g–m** (Figure 3). In addition to cyclic *N*-silylamines, acyclic substrates such as *N,N*-dimethyl- and *N,N*-diethyl-*N*-(trimethylsilyl)amine participated in the reaction to afford 3-(dialkylamino)-2-silylaryl triflates **1g** and **1h**, respectively, in good yields. *N*-Silylamines bearing an ethoxycarbonyl or *N*-(*tert*-butoxycarbonyl)amino group tolerated this transformation, as demonstrated in the synthesis of **1i** and **1j**. The reaction with *N*-aryl-*N*-methyl-*N*-(trimethylsilyl)amines, including those bearing an electron-donating or electron-withdrawing group on the aryl group, also took place to provide (*N*-methylanilino)aryne precursors **1k–m** in moderate yields.

3-Aminobenzene reacted with a broad range of arynophiles with high regioselectivities in a fashion similar to that of 3-alkoxybenzene (Table 2). For example, on treating 3-morpholinobenzene precursor **1a** with potassium fluoride and 18-crown-6-ether in the presence of *N*-methylaniline (**8**), an unsymmetrical 1,3-diaminobenzene derivative **9** was obtained as a single regioisomer in high yield (entry 1). Similarly, regioselective nucleophilic addition of thiol **10** to 3-morpholinobenzene took place efficiently to yield 3-morpholinophenyl sulfide **11** (entry 2). The cycloaddition of 3-morpholinobenzene with nitron **12** proceeded smoothly to

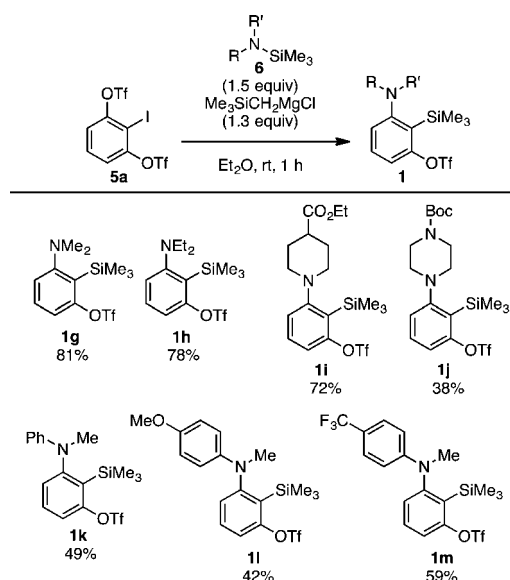


Figure 3. Synthesis of various 3-amino-2-silylphenyl triflates.

Table 2. Reactions of 3-Aminobenzene with Various Arynophiles

entry	arynophile	product ^a	yield (%)
1			84
2			94
3			96 (95:5) ^b
4			68
5			76
6			85
7 ^c			66

^aN = morpholino. ^bRatio of regioisomers, determined on the basis of ¹H NMR analysis, shown in parentheses. ^cThe reaction was performed using 2.0 equiv of **20** at 60 °C for 15 h.

afford the cycloadduct **13** with high selectivity (entry 3). When reactions were performed using ketene dimethyl acetal (**14**) or benzyl azide (**16**) as an arynophile, the corresponding cycloadducts **15** and **17** were exclusively obtained in good yields via [2 + 2] and [2 + 3] cycloadditions, respectively (entries 4 and 5). Diels–Alder reaction between 3-morpholinobenzene and 2,5-dimethylfuran (**18**) also proceeded efficiently to give naphthalene derivative **19** (entry 6). Furthermore, similar to our recent report regarding the direct thioamination of 3-methoxybenzynes with sulfilimine **20**,¹¹ 3-morpholinobenzynes reacted regioselectively with **20** to afford 1,3-diaminophenyl sulfide **21** as a single product in good yield (entry 7).

3-Aminobenzynes were also useful for preparing a small red-shifted fluorescent molecule¹² (Scheme 2 and Table 3).

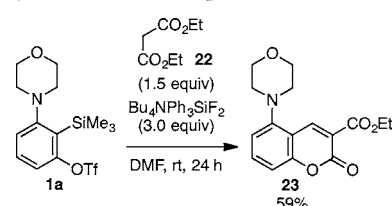
Scheme 2. Synthesis of 5-Morpholinocoumarin **23**

Table 3. Fluorescent Properties of Aminocoumarins

coumarin				
	23		24	
solvent	CH ₂ Cl ₂	MeOH	CH ₂ Cl ₂	MeOH
λ _{max} abs ^a (nm)	378	383	418	419
λ _{max} em ^b (nm)	540	573	452	465
Stokes shift (nm)	162	190	34	46
color				
	yellow	orange	blue	blue

^aWavelength of maximum absorption. ^bWavelength of maximum fluorescent intensity.

Although various 7-aminocoumarins have been prepared and their fluorescent properties have been studied, the fluorescent properties of 5-aminocoumarins remain largely unexplored, which could probably be attributed to the lack of a general synthetic method. Applying the reported method for aryne-mediated synthesis of coumarins¹³ to 3-aminobenzynes, we successfully prepared 5-morpholinocoumarin **23**; three-component coupling of 3-morpholinobenzynes precursor **1a**, diethyl malonate (**22**), and *N,N*-dimethylformamide resulted in the formation of **23** in good yield (Scheme 2). A solution of newly synthesized 5-morpholinocoumarin **23** in methanol or dichloromethane showed distinctive fluorescent properties as compared to that of 7-aminocoumarin such as **24**. Remarkable red-shifted emissions (from blue to yellow or orange) with dramatically expanded Stokes shifts (162 nm in dichloromethane and 190 nm in methanol) were observed. Although

further studies are required, the unique fluorescent properties of 5-morpholinocoumarin **23** can be explained by a twisted-intramolecular charge transfer (TICT) process,¹⁴ which is caused by the repulsion between the morpholino group and the peri hydrogen at the 3-position.

In summary, we have demonstrated the utility of an aryne relay chemistry for the synthesis of diverse aminoarenes. Various 3-amino-2-silylaryl triflates were easily prepared via silylation of 3-(triflyloxy)arynes. The key to the success was generation of 3-(triflyloxy)arynes via an iodine–magnesium exchange reaction of readily available 1,3-bis(triflyloxy)-2-iodoarenes using a silylmethyl Grignard reagent, which allowed the regioselective reaction with *N*-silylamines without cleaving their weak N–Si bond. A wide variety of aromatic amines, which are difficult to synthesize using conventional methods, have been efficiently synthesized with high regioselectivities from a 3-aminobenzene precursor prepared by our new method. Further studies expanding the scope of *N*-silylamines and the synthesis of other 5-aminocoumarin derivatives are now in progress.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b03304.

Experimental procedures, characterization for new compounds including NMR spectra (PDF)

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Notes

The authors declare no competing financial interest.

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